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EXAMINER

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**Please find below and/or attached an Office communication concerning this application or proceeding.**

The time period for reply, if any, is set in the attached communication.



**ELECTRODE FOR SECONDARY BATTERY, PROCESS OF PRODUCING THE  
ELECTRODE, AND SECONDARY BATTERY**

Examiner: K. Han    SN: 10/522,311    Art Unit: 1795    September 1, 2009

**Detailed Action**

1.     The Applicant's amendment filed on May 4, 2009 was received. Claims 13, 15, 22, 25, and 26 were amended. Claim 27 was added.
2.     The text of those sections of Title 35, U.S. Code not included in this action can be found in a prior Office action.

***Claim Objections***

3.     The objection to claims 13 and 25 are withdrawn in view of Applicant's comments.

***Claim Rejections - 35 USC § 112***

4.     The claim rejections under 35 U.S.C. 112, second paragraph, on claims 15, 22, 25, and 26 are withdrawn, because the claim has been amended.

***Claim Rejections - 35 USC § 102***

5.     Claims 13, 14, and 17-27 are rejected under 35 U.S.C. 102(b) as being anticipated by Kawakami et al. (US 6051340).

Regarding claim 13, Kawakami is directed towards an electrode for a secondary battery [Abstract] comprised of the following:

Art Unit: 1795

- a first and a second surface (Figure 4(a) and 4(d)) both with electrical conductivity and brought into contact with an electrolytic solution (Column 11, Lines 38-53),
- an active material layer containing a powdery material (103) positioned between the first and second surface (Column 10, Lines 6-33; Figures 4(c) and Figure 4(d)),
- the powdery material containing alloyable metal (active material) and non-alloyable material (electrically conductive metallic material) mixed together (Column 13, Lines 50-66),
- an electrically metal layer (101) in the middle of the thickness direction, and
- the active material is present on both sides of the conductive foil (Figure 4(d); Column 10, Lines 25-33).

Regarding claim 14, Kawakami discloses an example of an anode electrode having a total thickness of between 50 to 110 $\mu$ m (Columns 21-24).

Regarding claim 17, Kawakami discloses the electrode as an anode [Abstract].

Regarding claims 18 and 19, Kawakami discloses the surface of the anode to be covered by an insulating film (Column 16, Line 23-26) which allows the lithium ions to pass. The layers are formed from a powdery material formed from a sintered body (Column 5, Lines 22-30) which would inherently have a porous structure (microvoids) allowing the electrolyte to pass.

Regarding claim 20, Kawakami discloses the formation of the layer comprised of the alloyable (active material particles) and nonalloyable metal with electrically conductive auxiliary to form the anode which has a current collecting function as a whole (Column 13, Line 53-Column 14, Line 36).

Regarding claim 21, Kawakami teaches examples of the electrode having a total thickness as discussed for claim 14. The insulating film is disclosing as a surface layer formed by an insulating film (Column 16, Line 23-26) which has the property of allowing lithium ion to pass but prohibiting lithium metal. This film would inherently have a fractional thickness within the total thickness of the electrode.

Regarding claim 22, Kawakami discloses active material having a metal capable of being alloyed with lithium (103) (Column 10, Lines 9-10).

Regarding claims 23 and 24, Kawakami discloses forming the active material layer by obtaining a paste (slurry) and surface layers by electroplating (Column 13, Line 11-Column 14, Line 3).

Regarding claim 25, the teachings of Kawakami as discussed above are herein incorporated. Kawakami further teaches a nonalloyable metal (material with "lower" capability of forming a lithium compound) in the active material layer (Column 13, Lines 39-46).

Regarding claim 26, Kawakami discloses a multilayer structure of the anode layer where the metal incapable of being alloyed with lithium is at an enhanced content at the anodes surface forming a multilayer surface in combination with the insulating film (Column 5, Lines 13-21; Figure 4d).

Regarding claim 27, Kawakami discloses the powdery material comprising the active material layer are mixed together (13:50-66) which would inherently have some degree of porosity (vacant spaces) due to the nature of the powder material.

***Claim Rejections - 35 USC § 103***

6. Claim 15 is rejected under 35 U.S.C. 103(a) as being unpatentable over Kawakami et al. as applied to claim 13 above and further in view of Beard (US 5147739).

Regarding claim 15, the teachings of Kawakami as discussed above are herein incorporated. Kawakami discloses active material having a metal capable of being alloyed with lithium (103) (Column 10, Lines 9-10) but is silent towards a metallic lithium layer provided between the conductive foil and the active material layer.

Beard teaches a lithium battery comprised of having an anode with a current collector 13, metallic lithium layer 14, and an active material layer 15 (Figure 1A) for the benefit of providing a electrochemical cell with the full voltage available from a pure lithium anode without the problems of dendritic growth or lithium cycling loss (Column 5, Line 25-Column 6, Line 6). It would have been obvious to one of ordinary skill in the art at the time of the invention have a metallic lithium layer between the conductive foil and active material layer of Kawakami because Beard teaches that this configuration provides a battery which has the full voltage available from a pure lithium electrode without the problems of dendritic growth or lithium cycling loss.

***Response to Arguments***

7. Applicant's arguments with respect to claims 13-15 and 17-27 have been considered but are moot in view of the new ground(s) of rejection.

***Conclusion***

8. Applicant's amendment necessitated the new ground(s) of rejection presented in this Office action. Accordingly, **THIS ACTION IS MADE FINAL**. See MPEP § 706.07(a). Applicant is reminded of the extension of time policy as set forth in 37 CFR 1.136(a).

A shortened statutory period for reply to this final action is set to expire THREE MONTHS from the mailing date of this action. In the event a first reply is filed within TWO MONTHS of the mailing date of this final action and the advisory action is not mailed until after the end of the THREE-MONTH shortened statutory period, then the shortened statutory period will expire on the date the advisory action is mailed, and any extension fee pursuant to 37 CFR 1.136(a) will be calculated from the mailing date of the advisory action. In no event, however, will the statutory period for reply expire later than SIX MONTHS from the date of this final action.

***Contact/Correspondence Information***

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Kwang Han whose telephone number is (571) 270-

Art Unit: 1795

5264. The examiner can normally be reached on Monday through Friday 8:00am to 5:00pm.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Dah-Wei Yuan can be reached on (571) 272-1295. The fax phone number for the organization where this application or proceeding is assigned is 571-273-8300.

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/K. H./  
Examiner, Art Unit 1795

/Dah-Wei D. Yuan/  
Supervisory Patent Examiner, Art Unit 1795